

This article was downloaded by:

On: 29 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Supramolecular Chemistry

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713649759>

2,4,6-Tris[4-(1-naphthyl)phenoxy]-1,3,5-triazine: formation of a unique piedfort-based host lattice with trigonal symmetry

Keith Henderson^a; David D. Macnicol^a; Paul R. Mallinson^a; Ian Vallance^a

^a Department of Chemistry, University of Glasgow, Glasgow, U. K.

To cite this Article Henderson, Keith , Macnicol, David D. , Mallinson, Paul R. and Vallance, Ian(1995) '2,4,6-Tris[4-(1-naphthyl)phenoxy]-1,3,5-triazine: formation of a unique piedfort-based host lattice with trigonal symmetry', *Supramolecular Chemistry*, 5: 4, 301 – 304

To link to this Article: DOI: 10.1080/10610279508233958

URL: <http://dx.doi.org/10.1080/10610279508233958>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

2,4,6-Tris[4-(1-naphthyl)phenoxy]-1,3,5-triazine: formation of a unique piedfort-based host lattice with trigonal symmetry

KEITH HENDERSON, DAVID D. MACNICOL*, PAUL R. MALLINSON and IAN VALLANCE

Department of Chemistry, University of Glasgow, Glasgow G12 8QQ, U.K.

(Received March 30, 1995)

X-Ray crystal structure analysis of the isopropanol clathrate of the title compound **1** has revealed a trigonal lattice, space group $R\bar{3}c$ with $a=24.760(3)$, $c=87.644(14)\text{\AA}$, $Z=48$ (host), in which self-assembly of host molecules, in the form of three crystallographically independent Piedfort units of symmetry C_{3v} , C_3 and D_3 located along c , has led to the formation of two types of trigonal cage, each occupied by an isopropanol guest molecule.

INTRODUCTION

As part of a systematic study of the rational design of new classes of crystalline inclusion compound¹, we have recently introduced the Piedfort concept² in which two tri-substituted aromatic molecules, arranged back-to-back, successfully mimic a single hexahost molecule³. In the 1,4-dioxane inclusion compound of 2,4,6-tris[4-(2-phenylpropan-2-yl)phenoxy]-1,3,5-triazine, previously reported², the Piedfort host unit has symmetry approximating to C_{3v} , though its exact symmetry is only C_3 . We now report results from a single-crystal X-ray analysis of the novel trigonal clathrate formed between 2,4,6-tris[4-(naphthyl)phenoxy]-1,3,5-triazine (**1**) and isopropanol.

EXPERIMENTAL

This inclusion compound was formed by recrystallisation of **1** from isopropanol at *ca.* 100°C in a sealed tube. The adduct, obtained as very squat hexagonal prisms, had a host-guest ratio of *ca.* 2:1 as determined by ¹HNMR analysis, and also by thermogravimetric analysis, TGA, which indicated tight retention of guest up to *ca.* 190°C, the main weight loss occurring in the range 220–240°C (N₂ atmosphere, heating rate 10°C/min). The

m.p. of the adduct, observed on the hot-stage microscope, is *ca.* 220–230°C. The preparation and initial spectroscopic characterisation of host **1** were as previously described².

Crystal data for C₅₁H₃₃N₃O₃· $\frac{1}{2}$ C₃H₈O. $M=765.4$, rhombohedral, space group $R\bar{3}c$, $a=24.760(3)$, $c=87.644(12)\text{\AA}$, $U=46540(12)\text{\AA}^3$, $Z=48(4/3)$ host molecules per asymmetric unit). $D_c=1.311\text{gcm}^{-3}$, $\lambda(\text{Mo-K}\alpha)=0.7107\text{\AA}$, $\mu=0.082\text{mm}^{-1}$. $T=123\text{K}$. Number of independent intensities 10488 from colourless hexagonal prism, $0.3\times 0.3\times 0.1\text{mm}$. $R=0.0749$, $R_w=0.2127$ for 4523 observed reflections ($I/\sigma(I)>2.0$). X-ray intensity measurements for all possible reflections with $\sin\theta/\lambda < 0.64\text{\AA}^{-1}$ were made by $2\theta-\omega$ scans on a Nonius CAD4 diffractometer. However, even at the experimental temperature of 123 K, the crystal scattered weakly. Only *ca.* 25 % of reflections with $\sin\theta/\lambda > 0.4\text{\AA}^{-1}$ were observed. The principal computer programs used in structure solution and refinement are listed in ref 4. Atomic coordinates, and selected bond lengths and angles are given in Tables 1 and 2. Phenyl rings were treated as rigid groups during refinement, but all other non-hydrogen atoms of the host were refined free. Hydrogen atoms were added by geometric construction. Difference maps indicated guest molecules on the crystallographic 3-fold axis, and a model of the guest was refined at full occupancy, in one of the two cavities. The second guest (in a cavity with site occupancy factor apparently less than unity) gave diffuse electron density, and was modelled in terms of two carbon atom positions.

RESULTS AND DISCUSSION

As illustrated in Fig. 1, (a)-(c), there are three different types of Piedfort unit comprising the host component in the isopropanol adduct of 2,4,6,-tris[4-(1-naphthyl)-

*To whom correspondence should be addressed.

Table 1 Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{Å}^2 \times 10^3$) for 1. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor

	x	y	z	$U(\text{eq})$
O(1A)	-1039(1)	27(1)	3610(1)	49(1)
N(1A)	-551(2)	-557(2)	3612(1)	37(1)
C(1A)	-523(2)	-13(2)	3612(1)	39(1)
C(2A)	-1624(1)	-511(1)	3623(1)	44(1)
C(3A)	-1842(1)	-783(1)	3765(1)	49(1)
C(4A)	-2454(1)	-1264(1)	3781(1)	48(1)
C(5A)	-2849(1)	-1471(1)	3655(1)	41(1)
C(6A)	-2632(1)	-1199(1)	3513(1)	61(2)
C(7A)	-2019(1)	-718(1)	3497(1)	55(1)
C(8A)	-3531(1)	-1968(1)	3677(1)	42(1)
C(9A)	-3875(1)	-1775(1)	3763(1)	49(1)
C(10A)	-4513(1)	-2168(2)	3781(1)	54(1)
C(11A)	-4807(3)	-2739(3)	3712(1)	63(2)
C(12A)	-4460(2)	-2941(2)	3622(1)	52(1)
C(17A)	-3808(2)	-2535(2)	3605(1)	40(1)
C(16A)	-3461(2)	-2758(2)	3527(1)	58(2)
C(15A)	-3751(2)	-3346(2)	3461(1)	70(2)
C(14A)	-4402(4)	-3737(3)	3481(1)	80(2)
C(13A)	-4742(3)	-3537(3)	3556(1)	76(2)
O(1B)	-744(1)	464(1)	4021(1)	50(1)
N(1B)	-637(2)	-377(2)	4019(1)	41(1)
C(1B)	-347(2)	247(2)	4020(1)	37(1)
C(2B)	-517(1)	1097(1)	4000(1)	42(1)
C(3B)	-340(1)	1498(1)	4125(1)	44(1)
C(4B)	-184(1)	2116(1)	4104(1)	47(1)
C(5B)	-206(1)	2332(1)	3959(1)	35(1)
C(6B)	-383(1)	1931(1)	3835(1)	41(1)
C(7B)	-539(1)	1313(1)	3855(1)	45(1)
C(8B)	-59(1)	3003(1)	3937(1)	39(1)
C(9B)	531(1)	3456(1)	3985(1)	45(1)
C(10B)	720(2)	4084(1)	3968(1)	52(1)
C(11B)	331(2)	4258(2)	3899(1)	51(1)
C(12B)	-270(2)	3807(2)	3849(1)	44(1)
C(17B)	-460(2)	3165(2)	3867(1)	37(1)
C(16B)	-1068(2)	2725(2)	3820(1)	44(1)
C(15B)	-1469(2)	2899(2)	3755(1)	50(1)
C(14B)	-1265(3)	3536(3)	3738(1)	53(1)
C(13B)	-688(3)	3972(2)	3783(1)	49(1)
O(1C)	635(2)	1220(2)	207(1)	57(1)
N(1C)	642(2)	295(2)	202(1)	53(1)
C(1C)	319(2)	592(2)	202(1)	54(1)
C(2C)	1272(1)	1532(1)	185(1)	48(1)
C(3C)	1512(1)	1709(1)	38(1)	45(1)
C(4C)	2153(1)	2018(1)	15(1)	45(1)
C(5C)	2554(1)	2150(1)	138(1)	41(1)
C(6C)	2314(1)	1973(1)	284(1)	46(1)
C(7C)	1673(1)	1664(1)	308(1)	48(1)
C(8C)	3253(1)	2466(1)	107(1)	38(1)
C(9C)	3620(1)	3019(1)	186(1)	49(1)
C(10C)	4260(1)	3354(1)	162(1)	52(1)
C(11C)	4530(2)	3141(3)	58(1)	53(1)
C(12C)	4167(2)	2586(2)	-24(1)	44(1)
C(17C)	3507(2)	2237(2)	2(1)	37(1)
C(16C)	3156(2)	1678(2)	-78(1)	41(1)
C(15C)	3428(2)	1458(2)	-181(1)	46(1C)
C(14C)	4073(2)	1796(3)	-204(1)	51(1)
C(13C)	4430(2)	2347(3)	-128(1)	51(1)
O(1D)	1153(1)	902(1)	2689(1)	47(1)
N(1D)	467(2)	-152(2)	2693(1)	35(1)
C(1D)	566(2)	424(2)	2694(1)	33(1)
C(2D)	1658(1)	801(1)	2674(1)	39(1)
C(3D)	1963(1)	949(1)	2534(1)	41(1)
C(4D)	2538(1)	982(1)	2520(1)	43(1)
C(5D)	2808(1)	866(1)	2645(1)	37(1)
C(6D)	2503(1)	718(1)	2785(1)	39(1)
C(7D)	1929(1)	686(1)	2799(1)	40(1)
C(8D)	3473(1)	978(1)	2628(1)	40(1)
C(9D)	3903(1)	1567(1)	2574(1)	44(1)
C(10D)	4527(1)	1735(2)	2560(1)	55(1)

Table 1 (Continued)

	x	y	z	U(eq)
C(11D)	4721(2)	1309(3)	2597(1)	54(1)
C(12D)	4284(2)	717(3)	2653(1)	45(1)
C(17D)	3649(2)	550(2)	2670(1)	39(1)
C(16D)	3213(2)	-58(2)	2720(1)	41(1)
C(15D)	3400(2)	-484(2)	2759(1)	56(1)
C(14D)	4036(3)	-307(3)	2742(1)	64(2)
C(13D)	4454(3)	269(3)	2693(1)	60(2)
CX1	-3519(4)	3393(4)	1510(1)	73(6)
CX2	-3821(4)	3462(4)	1378(1)	70(3)
CX3	-3650(8)	2713(5)	1524(1)	82(6)
CX4	-2839(4)	3888(7)	1508(1)	77(6)
CQ1	107(16)	76(15)	5584(1)	541(29)
CQ2	182(13)	-464(14)	5580(2)	207(14)

phenoxy]-1,3,5-triazine (**1**). All three Piedfortunits, located on the crystallographic *c*-axis, are constrained to have exact trigonal symmetry. Unit (b) has C_3 symmetry; whilst (a), located on a point of inversion, possesses C_{3i} symmetry; and unit (c), also located on the crystallographic two-fold axis at $z=\frac{1}{2}$, has exact D_3 symmetry. The relative dispositions of the two halves of units (a),(b), and (c) are described by the magnitudes of the respective

pseudo torsion angles, 60° , 21° and 24° , which correspond to the *c*-axis and the closest oxygen atoms in the adjacent triazines. The corresponding pseudo torsion angles for the closest oxygen atoms in *neighbouring* Piedfort units along *c* are 6° , between units (a) and (b), and 10° , between units (b) and (c). All the triazine rings are accurately planar; and for unit (b), atoms O(1A) and O(1B) are close to their respective triazine ring planes, magnitudes of displacements .013(5) and .015(9)Å, respectively. For units (a) and (b), however, atoms O(1C) and O(1D) are more displaced, by 0.045(6) and 0.037(5)Å, from their respective triazine ring planes. The bond lengths and angles closely parallel those reported previously² for 2,4,6-tris[4-(2-phenylprop-2-yl)phenoxy]-1,3,5-triazine. Interestingly, the N-C-O angles for a given oxygen atom are markedly unequal, mean angles corresponding respectively to *cisoid* and *transoid* side-chain dispositions being 119.6° and 112.6° . The inter-triazine plane distances for Piedfort units (a),(b), and (c) are 3.544(8), 3.574(5), and 3.390(6)Å, respectively.

The intricate host-guest packing in the isopropanol adduct of **1** is illustrated in Fig. 2. There are two types of crystallographically distinct void, and these are located along the *c*-axial direction which is vertical in the view shown. Both cavities have exact C_3 symmetry and each is occupied by an isopropanol guest molecule, although one of these is highly disordered and could not be located unambiguously. Only the better ordered isopropanol guest is shown in Fig. 2. The ends of the cavity containing the better ordered guest are bounded by Piedfort units, (b) and (c), with C_3 and D_3 symmetry, whilst its walls are formed by three symmetry related C_{3i} , (a) units and three C_3 , (b) units. The cavity containing the highly disordered guest has ends comprised of (a) and (b) Piedfort units, and walls comprised of three (b) units and three (c) units. Computed van der Waals sections at 1Å spacing along *c* show that both cavities are of the completely closed cage type (in keeping with the extremely tight guest retention found by TGA), estab-

Table 2 Selected bond lengths [Å] and angles [deg] for **1**

O(1A)-C(1A)	1.331(5)
O(1A)-C(2A)	1.397(3)
N(1A)-C(1A)	1.313(5)
N(1A)-C(1A)#1	1.343(5)
O(1B)-C(1B)	1.335(5)
O(1B)-C(2B)	1.388(3)
N(1B)-C(1B)#1	1.319(6)
N(1B)-C(1B)	1.340(5)
O(1C)-C(1C)	1.347(5)
O(1C)-C(2C)	1.381(4)
N(1C)-C(1C)#2	1.325(6)
N(1C)-C(1C)	1.331(6)
O(1D)-C(1D)	1.340(5)
O(1D)-C(2D)	1.401(3)
N(1D)-C(1D)	1.320(5)
N(1D)-C(1D)#2	1.334(5)
C(1A)-O(1A)-C(2A)	120.3(3)
C(1A)-N(1A)-C(1A)#1	113.0(5)
N(1A)-C(1A)-O(1A)	121.0(4)
N(1A)-C(1A)-N(1A)#2	127.0(5)
O(1A)-C(1A)-N(1A)#2	112.0(4)
C(1B)-O(1B)-C(2B)	119.4(3)
C(1B)#1-N(1B)-C(1B)	112.9(4)
N(1B)#2-C(1B)-O(1B)	120.2(4)
N(1B)#2-C(1B)-N(1B)	127.1(4)
O(1B)-C(1B)-N(1B)	112.7(4)
C(1C)-O(1C)-C(2C)	118.9(4)
C(1C)#2-N(1C)-C(1C)	111.9(5)
N(1C)#1-C(1C)-N(1C)	128.1(5)
N(1C)#1-C(1C)-O(1C)	113.5(5)
N(1C)-C(1C)-O(1C)	118.4(5)
C(1D)-O(1D)-C(2D)	121.1(3)
C(1D)-N(1D)-C(1D)#2	111.1(4)
N(1D)-C(1D)-N(1D)#1	128.9(4)
N(1D)-C(1D)-O(1D)	119.3(4)
N(1D)#1-C(1D)-O(1D)	111.8(4)

Symmetry transformations used to generate equivalent atoms: #1 -y,x-y,z #2 -x+y, -x,z

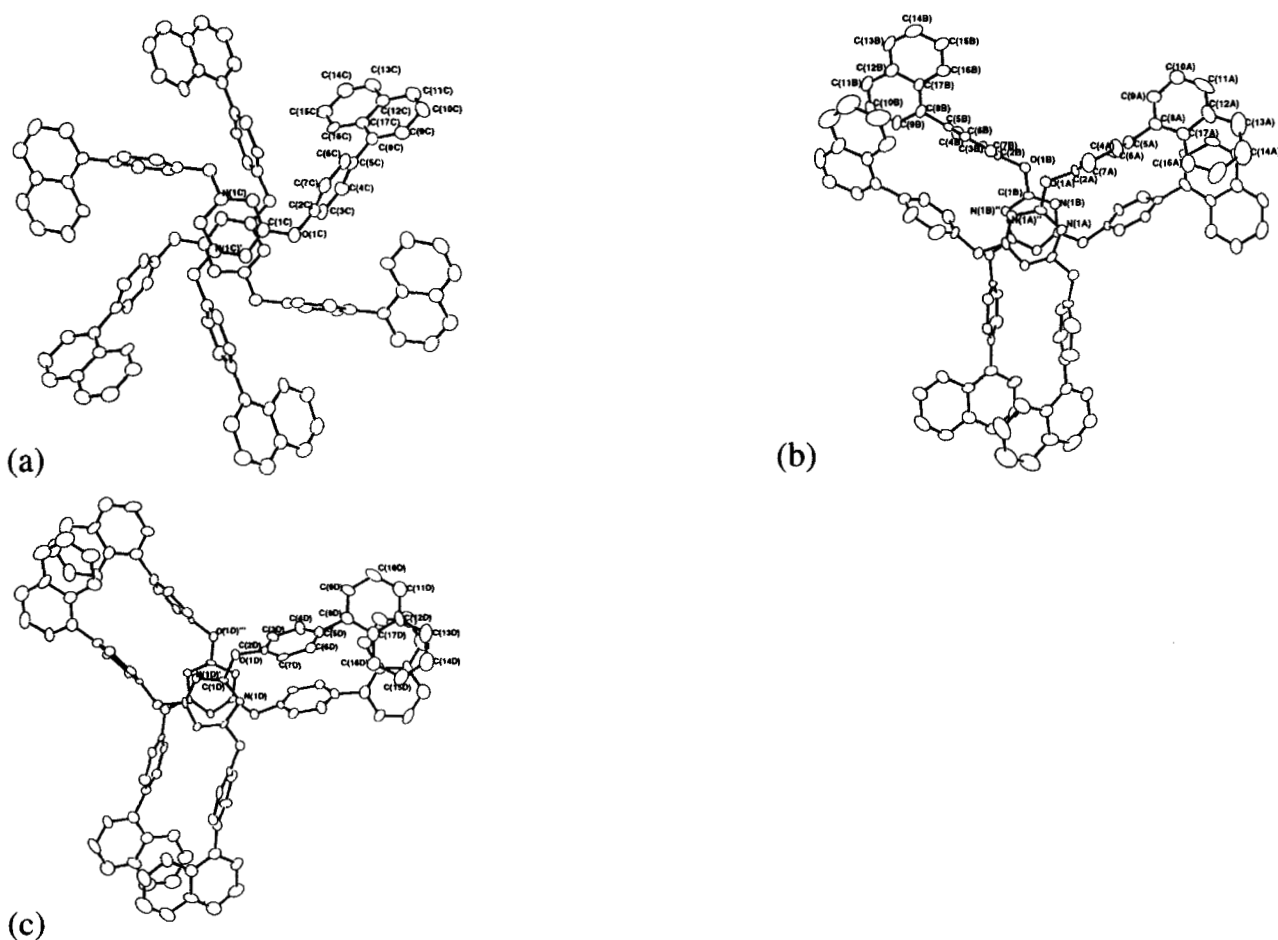


Figure 1 An illustration of the three distinct types of Piedfort unit present in the isopropanol clathrate of 2,4,6-tris[4-(1-naphthyl)phenoxy]-1,3,5-triazine **1**; the three units, viewed at a narrow angle to the crystallographic c -axis, have exact respective symmetries C_{3i} , C_3 , and D_3 . The disposition of the side-chain in unit (a) is described by torsion angles $N(1C)-C(1C)-O(1C)-C(2C)$, $-11.5(7)^\circ$, and $C(1C)-O(1C)-C(2C)-C(7C)$, $89.2(4)^\circ$; whilst corresponding torsion angles for unit(b) are $N(1A)-C(1A)-O(1A)-C(2A)$, and $N(1B)-C(1B)-O(1B)-C(2B)$, $6.1(7)^\circ$ and $9.0(6)^\circ$; $C(1A)-O(1A)-C(2A)-C(7A)$, and $C(1B)-O(1B)-C(2B)-C(7B)$, $-110.0(7)^\circ$ and $96.6(4)^\circ$; and for unit(c), $N(1D)-C(1D)-O(1D)-C(2D)$, and $C(1D)-O(1D)-C(2D)-C(7D)$, are $4.4(6)^\circ$ and $-85.7(4)^\circ$, respectively.

lishing the true classical clathrate character of this adduct of **1**. The potential storage of gaseous guest species in Piedfort-based rhombohedral host lattices of this type is currently under consideration.

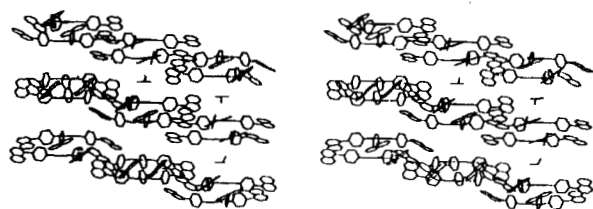


Figure 2 A stereoview, looking normal to the c -axis, showing the host-guest packing in the isopropanol clathrate of **1**. There are two types of crystallographically independent trigonal cage; each is occupied by an isopropanol guest molecule, but only the better ordered guest (representative orientation) is shown.

REFERENCES

- 1 See, for example, D.D. MacNicol in *Inclusion Compounds*, eds. J.L. Atwood, J.E.D. Davies and D.D. MacNicol, Academic Press, London 1984, vol. 2., ch. 5.
- 2 A.S. Jessiman; D.D. MacNicol; P.R. Mallinson; I. Vallance; *J. Chem. Soc., Chem. Commun.* **1990**, 1619–1621.
- 3 A.D.U. Hardy; D.D. MacNicol; D.R. Wilson; *J. Chem. Soc., Perkin Trans. 2* **1979**, 1011–1019; D.D. MacNicol; D.R. Wilson; *J. Chem. Soc., Chem. Commun.* **1976**, 494–495.
- 4 G.M. Sheldrick (1985), SHELXS86, Program for Crystal Structure Determination. Univ. of Göttingen, Germany; G.M. Sheldrick (1994), SHELXL93, Program for Crystal Structure Refinement, Univ. of Göttingen, Germany